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\_\_\_ 1. The first step in most stoichiometry problems is to \_\_\_\_. a. add the coefficients of the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities to masses \_\_\_ 2.

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Map HW: 10.2 Section Assessment 11  
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Chapter 12 Assessment. SE, p. 369 TWE, p. 368 SE, pp. 378–383 5 minutes [total 90 minutes]

## **LESSON PLAN 12 - glencoe.com**

d. 12% e. 2% 11. How many grams of H<sub>2</sub>O will be formed when 32.0 g H<sub>2</sub> is allowed to react with 16.0 g O<sub>2</sub> according to  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  a. 9.00 g b. 16.0 g c. 18.0 g ... Practice Test Ch3 Stoichiometry (page 3 of 3) 1. d It might be easiest to balance the equation with mostly whole numbers:  $2\text{NH}_3 + \frac{7}{2}\text{O}_2 \rightarrow 2\text{NO}_2 + 3\text{H}_2\text{O}$ . The ...

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