

## Gpb Episode 803 Limiting Reagents Answer Key

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atomic mass unit - equals 1/12 the mass of a carbon atom. Avagadro's number - the number of atoms in a mole, equal to  $6.02 \times 10^{23}$  atoms.. conversion factor - a ratio expressed as a fraction that equals one. dimensional analysis - the sequential application of conversion factors expressed as fractions and arranged so that any dimensional unit can be cancelled out until the desired set of ...

## **Segment E: Limiting Reactants | Georgia Public Broadcasting**

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### **Chemistry & Physics | Chemistry 803: Limiting Reactants ...**

Excess Reactant \_\_\_\_\_ that is not used up in a chemical \_\_\_\_\_ Ex. Problem: When  $\text{FeCl}_3$  reacts with  $\text{O}_2$ ,  $\text{Fe}_2\text{O}_3$  and  $\text{Cl}_2$  are produced. If 4.0 moles of  $\text{FeCl}_3$  and 4.0 moles of  $\text{O}_2$  are mixed, how many grams of  $\text{Fe}_2\text{O}_3$  will be produced? (Hint: Work two separate problems, using one reactant at a time.)

### **Stoichiometry Problem Guidelines: 1. the equation 2 ...**

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